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Chapter 16 Thermal Energy and Heat. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Sokollm (In order of which they appear) Key Concepts: Terms in this set (20) Heat. the transfer of thermal energy from one object to another because of a difference in temperature. Temperature.

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a measure of how hot or cold an ...

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Physical Science Chapter 16: Thermal Energy and Heat. Heat is the transfer of thermal energy from one object to another as the result of a difference in _____ . _____ produces heat.

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It states that thermal energy can flow

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from colder objects to hotter objects only if work is done on the system. Third law of thermodynamics It states that absolute zero cannot be reached.

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volume of material when its temperature increases. Conduction. is the transfer of thermal energy through touching with no ove.... Heat Engine. is any device that converts heat into work. 36 Terms. etakp. Chapter 16 Thermal Energy and Heat.

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Chapter 16: Thermal Energy and Heat
16.1 - Thermal Energy and Matter . Work and Heat Heat is the transfer of thermal energy from one object to another because of temperature differences Heat flows spontaneously from hot objects to cold objects Imagine two glasses with differing amounts of water in them at the same temperature The glass with ...

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Chapter 16 - Thermal Energy

16.1 Thermal Energy & Matter. Work and Heat. Heat -the transfer of thermal energy from one object to another because of a temperature difference Heat flows from higher temps to lower temps. Temperature is related to the kinetic energy of the particles: particles move around as they heat

Chapter 16

Chapter 16: Thermal Energy and Heat. Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; A B; heat: the transfer of thermal energy from one object to another

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because of a difference in temperature:
temperature: a measure of how hot or cold an object is compared to a reference point:

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Chapter 16 Thermal Energy and Heat
Section 161 Thermal Energy and Matter (pages 474-478) This section defines heat and describes how work, temperature, and thermal energy are related to heat Thermal expansion and contraction of materials is discussed, and uses of a calorimeter are explained
Reading Strategy (page 474)

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 $c = 0.128 \text{ J/g} \cdot \text{OC}$ Mass = $m = 25.0$ grams

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